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Journal of Research of the National Bureau of Standards

United States.
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1976

Chemistry John McMurry

1998 A full-year course taken primarily by chemistry majors, other science majors (especially biology and pre-health), and engineering students. First introduced in 1995, McMurry/Fay's Chemistry is now recognized as one of the leading books in science education. The second edition refines the qualities that led to the text's success in the first place. "The text is a beautifully presented and well written general chemistry text. The chapter on gas laws, where the combination of narrative and illustrations leads the students to almost derive the Kinetic-Molecular theory on their own." (Mildred Hall, Clark State Community College.)

Journal of the Royal

Institute of Chemistry Royal

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Material Balances for Chemical Reacting Systems R.L. Cerro
2022-12-05 Written for use in the first course of a typical chemical engineering program, *Material Balances for Chemical Reacting Systems* introduces and teaches students a rigorous approach to solving the types of macroscopic balance problems they will encounter as chemical engineers. This first course is generally taken after students have completed their studies of calculus and vector analysis, and these subjects are employed throughout this text. Since courses on ordinary differential equations and linear algebra are often taken simultaneously with the first chemical engineering course, these subjects are introduced as needed. Teaches readers the fundamental concepts

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associated with macroscopic balance analysis of multicomponent, reacting systems Offers a novel and scientifically correct approach to handling chemical reactions Includes an introductory approach to chemical kinetics Features many worked out problems, beginning with those that can be solved by hand and ending with those that benefit from the use of computer software This textbook is aimed at undergraduate chemical engineering students but can be used as a reference for graduate students and professional chemical engineers as well as readers from environmental engineering and bioengineering. The text features a solutions manual with detailed solutions for all problems, as well as PowerPoint lecture slides available to adopting professors.

General Chemistry for Engineers James O. Glanville 2002 Emphasizing problem-solving and engineering approximation, this chemistry

book provides engineers with an understanding of the entities (atoms, molecules, and ions) that are relevant to their lives and professional careers. Throughout the book, internet key word searching and graphing exercises take advantage of users' existing computer skills and encourages them to acquire new ones in designing, preparing, and interpreting graphs. Chapter topics cover atoms, elements, and measurements; nuclides, molecules, and ions; chemical reaction and stoichiometry; gases; quantum mechanics, and the periodic table; chemical bonding and chemical structure; chemical energy and the first law of thermodynamics; the second law of thermodynamics and chemical equilibrium; gas and solution equilibria; liquids and their mixtures; solids; phase diagrams and solutions; the periodic table and redox chemistry; electrochemistry; and rate processes. For engineers preparing for the professional certification exam.

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Addison-Wesley Chemistry

Antony C. Wilbraham 2001-02

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Current Research 1978

New Technical Books New

York Public Library 1952

British Books in Print 1968

Chemistry Eugene LeMay, Jr.

2002-02

Quimica: la Ciencia Central

Theodore L. Brown 2004 The scientific accuracy, clarity, and visuals of *Chemistry: The Central Science* make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text.

Translated into

Spanish. Introduction: Matter and Measurement, Atoms,

Molecules, and Ions,

Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

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Contributions Stanford University. Department of Chemistry 1955 Contains reprints of articles published by members of the department.

Basic Chemistry William S. Seese 1981

General Chemistry Ralph H. Petrucci 1972

Applied Mechanics Reviews
1987

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