

Acid And Base Test Review Sheet Answers

The pH scale ranges from 1 to 14 where 1 to 6 is classified as acidic, 7 neutral (neither a base or an acid) and 8 to 14 is classified as basic. In this lab, you will use the juice from red cabbage as a pH ... Name Perchloric acid Sulfuric acid Hydrochloric acid Nitric acid Trichloroacetic acid Hydronium ion Chromic acid Oxalic acid Nitrotriacetic acid Phosphoric acid EDTA Aquo ferric ion Arsenic acid ... Since H⁺ and any cation from a solvent autodissociation is a Lewis acid, and anything that can add H⁺ or a solvent-derived cation is a Lewis base, the Lewis acid concept effectively includes the ones ... A Brønsted-Lowry acid such as HCl is an acid-base adduct according to the Lewis concept, and proton transfer occurs because a more stable acid-base adduct is formed. Carboxylic Acids Ref. 77 Aminomalonic acid* 3.32, 9.83 N-Butylaminoacetic acid 2.29, 10.07 2-carboxyethyliminodiacetic acid 2.06, 3.69, 9.66 The names have the general form of hydro(root)ic acid, such as hydrochloric acid. The formulas have the general form of HX(aq) (where X is F, Cl, Br, or I) or H₂X(aq) (where X is S or Se) The formulas ... Conjugate acids (cations) of strong bases are ineffective bases. * Compiled from Appendix 5 Chem 1A, B, C Lab Manual and Zumdahl 6th Ed. The pKa values for organic acids can be found in Appendix II ...

[pKa Values INDEX - Organic Chemistry Data](#)

Formula Name pKa1 pKa2 pKa3 pKa4 4 -7 2 4 -3 3 -1

Three Types of Acids

Acid - Gelman Library

Acid-Base concepts - Texas A&M University

[Acids and Bases Edit - Stanford University](#)

CHAPTER 16: ACIDS AND BASES

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